

## 6 Solid HLW

Properties: activity\*, heat, thermal effects

Activity of radionuclides in high-level waste (Ci/canister)

Time out of Reactor - y

RN	Half-Life	10	100	1000	10000
90-Sr	29 y	1.4E+5	1.5E+4	3.5E-6	0
137-Cs	30.1 y	2.0E+5	2.4E+4	2.3E-5	0
154-Eu	8.6 y	1.0E+4	210	0	0
210-Pb	22.3 y	0	1.7E-6	1.6E-4	6.5E-3
226-Ra	1600 y	2.5E-7	2.6E-6	1.6E-4	6.5E-3
229-Th	7340 y	9.6E-8	1.7E-6	1.6E-4	1.3E-2
230-Th	7.7E+4 y	4.9E-5	7.4E-5	8.8E-4	8.1E-3
238-Pu	87.8 y	2.4E+2	120	0.28	0
239-Pu	2439 y	3.8	3.8	4.7	9.6
240-Pu	6540 y	10	20	18	7.3
241-Pu	15 y	740	11	0.72	0.34
241-Am	433 y	410	380	81	0.34
243-Am	7370 y	40	40	37	17
242-Cm	163 d	16	11	0.18	0
244-Cm	17.9 y	4.0E+3	130	0	0

Short-lived RI at 10 y: 106-Rh 369 d 1300; 125-Sb 2.73 y 1600; 134-Cs 2.06 y 1.9E+4; 144-Ce 284 d 35; 147-Pm 2.62 y 1.8E+4.

Long-lived RI at 10000 y : 93-Zr 9.5E+5 y 4.3, 93m-Nb 12 y 4.3,

99-Tc 2.1E+5 y 32, 126-Sn 1E+5 y 1.3, 129-I 1.6E+7 y 1.8E-4, 233-U 1.6E+5 y 3.6E-2, 237-Np 2.1E+6 y 0.88

**Contribution to heat generation of thermally significant isotopes in PWR spent fuel (%).**

time - d	30	90	365	1096	3650
Ba/La-140		13.8	0.93		
Zr/Nb-95		23.2	24.5	3.9	
Ce/Pr-144		18.0	26.9	35.3	17.3
Cm-242	2.2	3.0	2.36		
Cs/Ba-137	1.1	2.0	5.0	14.1	35.3
Sr/Y-90	1.1	2.0	4.9	13.7	33.8
Pu-238		0.33	0.85	2.5	6.9
Cm-244			0.74	2.0	4.5

Phase separation: unstable, T + t : Na<sub>2</sub>MoO<sub>4</sub> (Sr,Cs) soluble, MoO<sub>3</sub> lesssoluble, Li<sub>2</sub>O, 15% B<sub>2</sub>O<sub>3</sub>; crystallization: 400-800 C, nucleation, crystal growth, leachability 1000X, waste loading 50-100 kW/m<sup>3</sup>, container 0.3 m, T grad 500 C, 1960, P glass 455 C 120 d leaching 100X; BSi glass, devitrification products: Zn<sub>2</sub>SiO<sub>4</sub>, (CaSrBa)MoO<sub>4</sub>, apatite, cracking, porosity +3%, no devitrification if T<450 C;

Waste forms: calcines, glass, glass-ceramics, synthetic minerals

**Characteristics of world's solidified high-level wastes.**

Solidification process	P	WL	HTF	Density	TC	LL
calcination US	C	100	500	1.0-1.7	0.4-1.0	5E-1
calcination ORNL US	C	90	900	1.2-1.4	0.6-1.0	5E-1
ORNL/WSEP US	BG	30-50	900	2.9-3.1	-	1E-6
WSEP US	G	20-40	1200	3.0	-	1E-4

P GLASS	BNL	US	PG	30	1200	2.7-2.9	2.0-3.5	1E-6
STOPPER		US	S	-	500	-	-	1E-8
THERMALT		US	AS	-	2000	2.9	-	1E-8
FINGAL	AERE	UK	BG	25-40	1050	2.8	2.5-4.0	1E-7
HARVEST	AERE	UK	BG	25	1050	2.6	2.8-4.0	1E-7
Dis. process	CEA	F	BG	20-30	1150	-	2.8-3.6	1E-7
Cont. process	CEA	F	BG	20-30	1150	-	2.8-3.6	1E-7
VERA	Germany		BG	20-30	1200	2.5-2.7	-	1E-7
PHOTO	Germany		PG	25-35	1100	2.6-2.9	2.4	1E-7
LOTES	Eurochemic		PG	30	300	2.1	3.0-4.0	1E-5
Pot solidif.	India		BG	22-28	1050	2.5-3.0	2.5-3.5	1E-6
ESTER	Italy		BG	20-25	1000	2.7-3.0	-	1E-7
ESTER	Italy		PG	20-25	900	2.3-3.5	-	1E-5

F = France, P = Product (A = alumino, G = glass, B = borosilicate, P = phosphate, S = silicate), WL = waste loading %, HTF = Highest Temperature of Formation, TC = Thermal Conductivity in mcal/cm/s/C, LL = Lowest Leachability in g/cm<sup>2</sup>/d.

Calcines: early form, oxides, solubles

Glass: thermodynamically unstable, supercooled liquid +: easy, known, \$, durable, variable composition 20% FP -: fragile, unstable; structure: O polyhedra, attraction O-cation,

repulsion Cat-Cat;

Ionic field strengths (FS :  $Z/r^2$ ), charge (C) and ionic radius (IR in  $1E-12$  m) of cations present in glasses.

	network formers				intermediate ions				network modifiers			
	B	P	Si	As	Be	Al	Ti	Zr	Mg	Li	Ca	Na
C	3+	5+	4+	5+	2+	3+	4+	4+	2+	1+	2+	1+
IR	20	34	41	47	31	50	68	80	65	60	99	95
FS	75	43	24	23	21	12	9	6.3	4.7	2.8	2.0	1.

Oxygen ratio (OR=O/Si): OR = 2 ( $SiO_2$ ), glass:  $2 < OR < 3$   
 OR = 4 no glass; leaching:  $Na^+$ ,  $H^+$  diffusion and exchange,  
 alkali deficient layer, kinetics:  $t^{0.5}$ , linear: long t, high T;  
 glass-ceramic: dispersion of crystals, annealing at nucleation  
 T + GG, celsian, diopside, eucryptite, perovskite,  
 borosilicate G

SYNROC and its constituent minerals (weight %, H= Hollandite, Z = Zirconolite, P = Perovskite, BS = Bulk SYNROC).

	H	Z	P	BS
TiO <sub>2</sub>	71.0	50.3	57.8	60.3
ZrO <sub>2</sub>	0.2	30.5	0.2	10.8
Al <sub>2</sub> O <sub>3</sub>	12.9	2.5	1.2	6.3
CaO	0.4	16.8	40.6	16.2
BaO	16.0	-	-	6.4

properties: stable, 20 % RW, resistance (mechanical, chemical, radiation); Distribution of high-level waste elements among constituent minerals of SYNROC.

Hollandite : Cs<sup>+</sup>, Rb<sup>+</sup>, K<sup>+</sup>, Ba<sup>2+</sup>, Fe<sup>3+</sup>, Cr<sup>3+</sup>, Ni<sup>2+</sup>, Mo<sup>4+</sup>

Zirconolite : U<sup>4+</sup>, Th<sup>4+</sup>, Pu<sup>4+</sup>, Cm<sup>4+</sup>, Np<sup>4+</sup>, Act<sup>3+</sup>, RE, Sr<sup>2+</sup>

Perovskite : Na<sup>+</sup>, Sr<sup>2+</sup>, Pu<sup>3+</sup>, Cm<sup>3+</sup>, Np<sup>3+</sup>, RE, Act<sup>4+</sup>

Metal : Ru, Tc, Mo, Ni, Pd, Rh, Te, S, Fe

leaching: removal of C+, C++, Ti Zr rich film, Cs, Ba 1000X, Nd<sup>3+</sup>, U<sup>4+</sup> 10000X less than glass, no new phases; radiation damage: natural Z + P alpha emitters, 1.3E+18 a/g (1000 y) dV +2%, 8E+18 (400 000 y) +3%, Z monoclinic to cubic (CaF<sub>2</sub>) structure, 8E+19 a/g (4E+8 y) metamict +3% crystalline state, P 20000 y +1.8%; Sri Lanka Z 1E+18 a/g 200 C bulk leach 3E-4 g/m<sup>2</sup>/d, 8E+18 a/g 2E-3 g/m<sup>2</sup>/d; preparation: precipitation TIPTi + TBZr + nitrates, slurry, drying 130 C powder, calcination 750 C Ar 4% H<sub>2</sub> Cs Ru, HIP ~1150 C ~ 125 MPa, uniaxial HP, 4.5 g/cm<sup>3</sup> more waste, clean; disposal: boreholes 1 m, 4 km, waste : 2.5 km,

seal: 1.5 km 80 GWy, drilling damage, sealing

Interim storage: wet, dry, fuel bays 40 y, 67 US plants "full core reserve" by 2000, dry storage, AFR;

Wet S\*: 95% pools 12 m deep, 3 m shielding +: cooling -: dirty; capacity increase: reracking, poisoned rack B, rod consolidation, dismantling FA, storage canister, old fuel

Dry S: used for 20 y by DOE, +: flexibility, passive cooling, cleaner, metal casks, concrete silos\*\*, dry wells (Nevada), buildings\*

Large metal dry storage cask characteristics

Characteristic	Castor TN-24P MC-10* NAC 5100			
NRC license status	L	U	U	U
Capacity (PWR)	21	24	24	28
Construction materials	iron	steel	steel	SS/Pb/SS
Neutron shielding	PE	BP	BP	SS/Pb/SS
Maximum weight-tons	115	100	100	?

L = Licensed, U = Unlicensed, SS = Stainless steel, PE = Polyethylene, BP = Borated plastic.

AFR: 2 wet facilities Morris (Ill), West Valley (NY), dry AFR Gorleben (G)

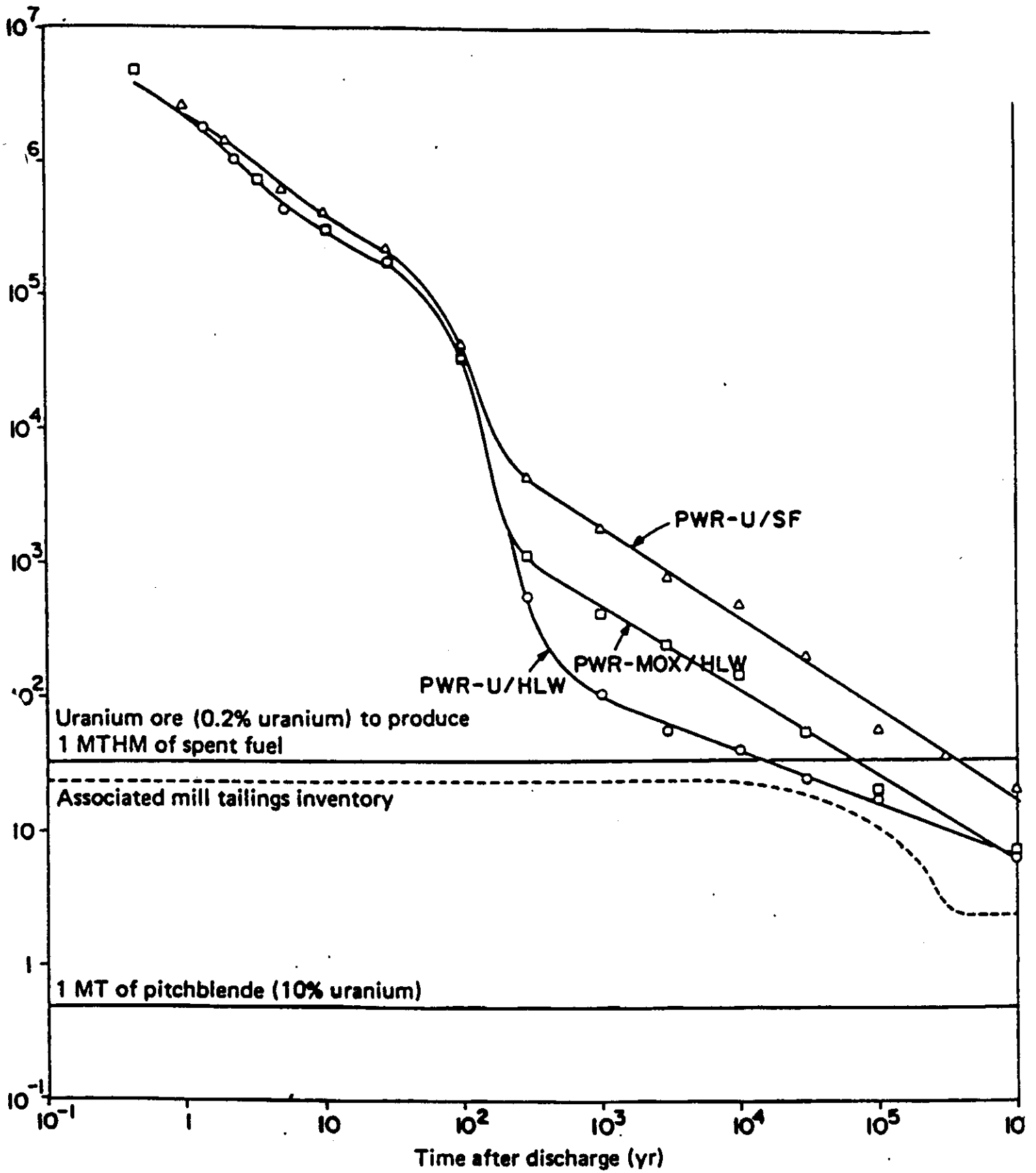


Fig. 1. Radioactive decay of PWR spent fuel and HLW.

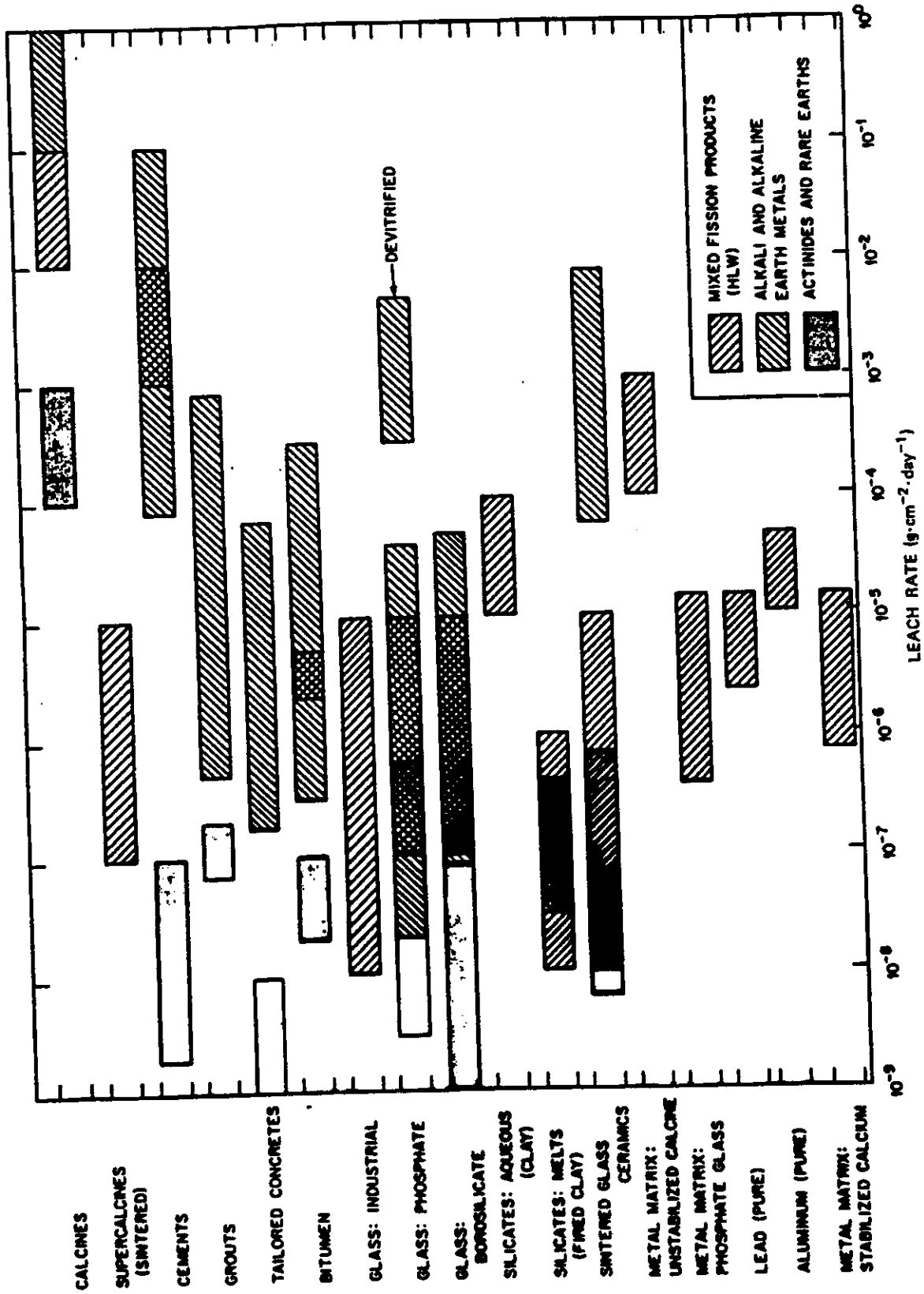
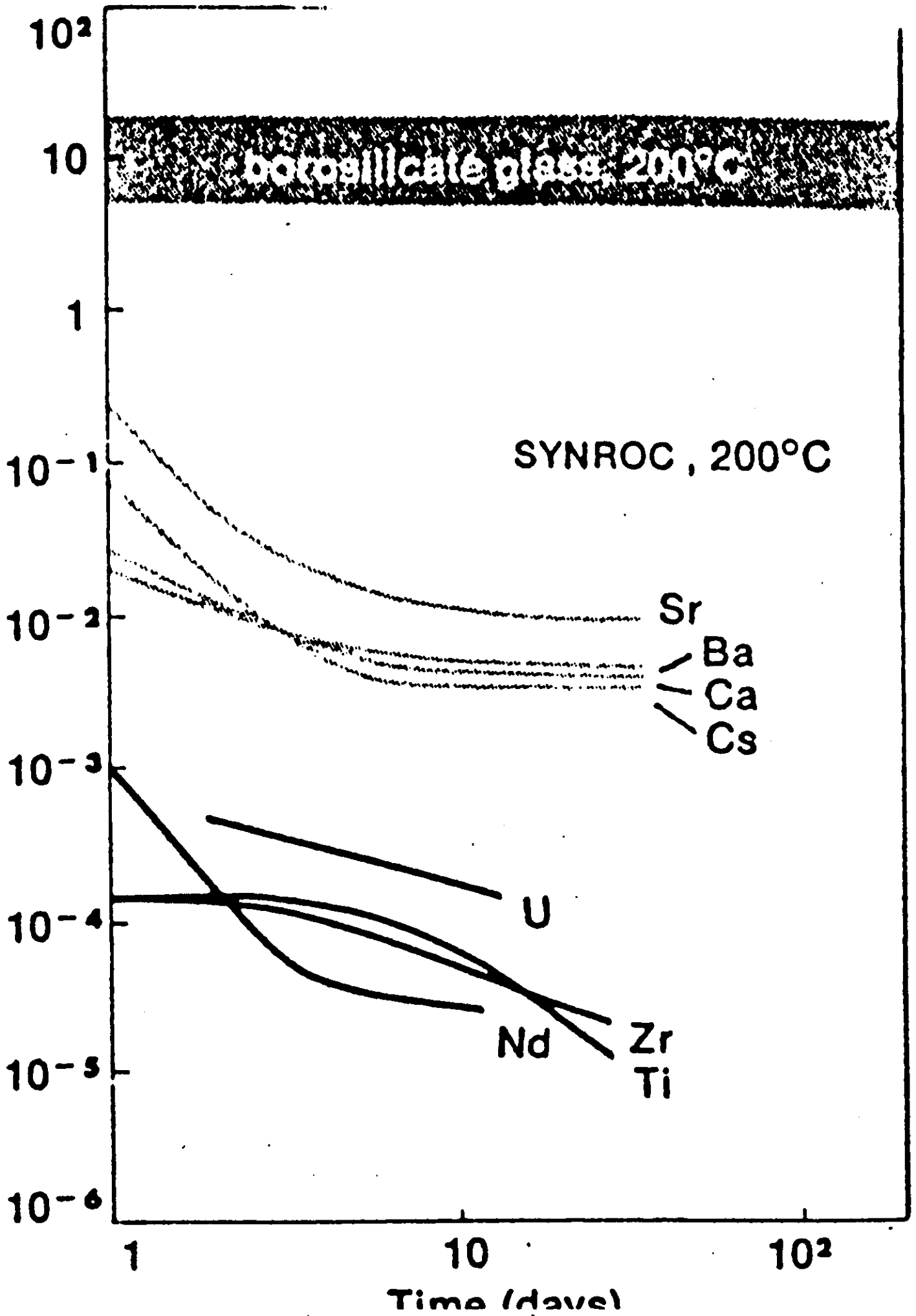


Fig. 5. Some reported leach rates for various radionuclides in several matrix materials.





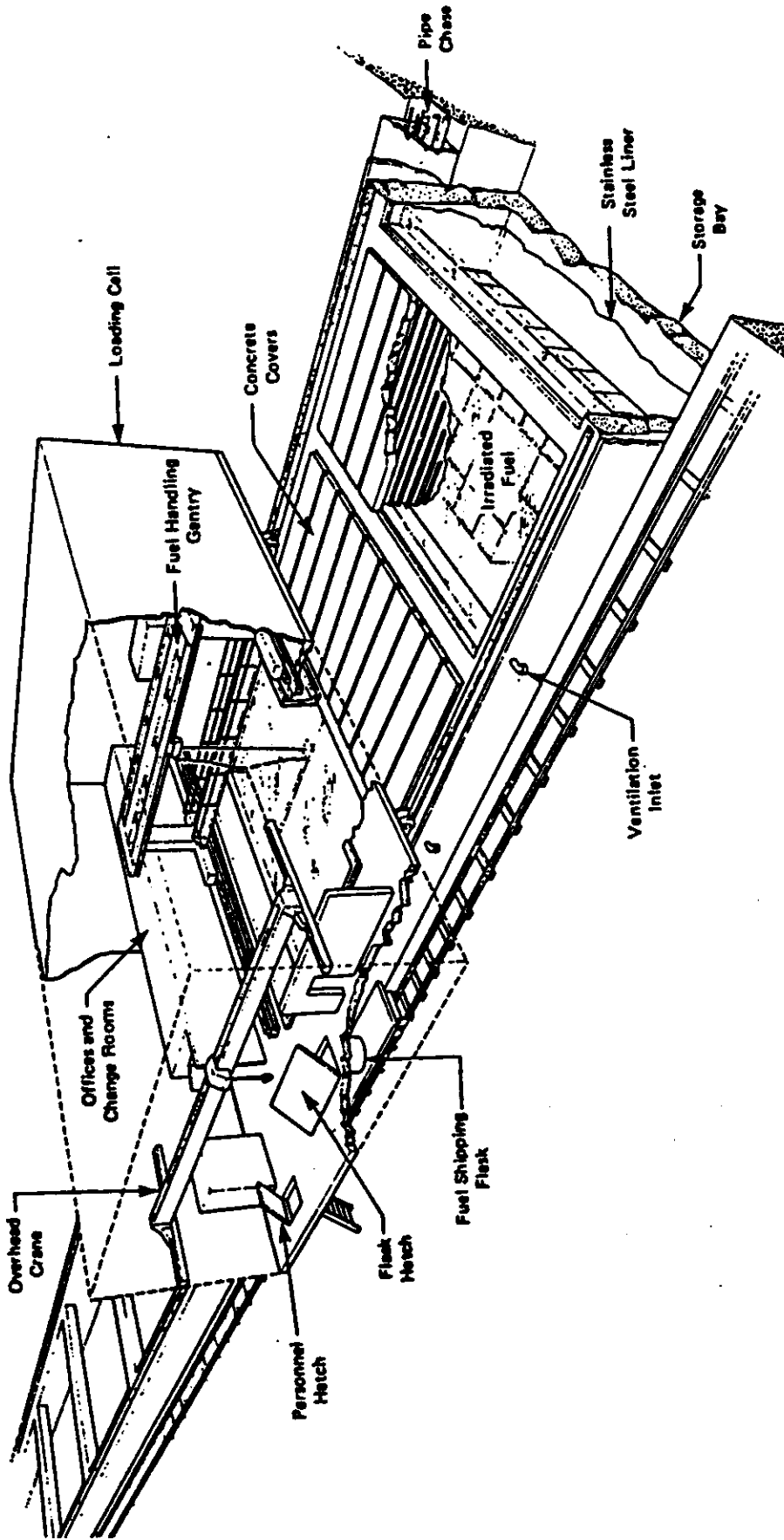
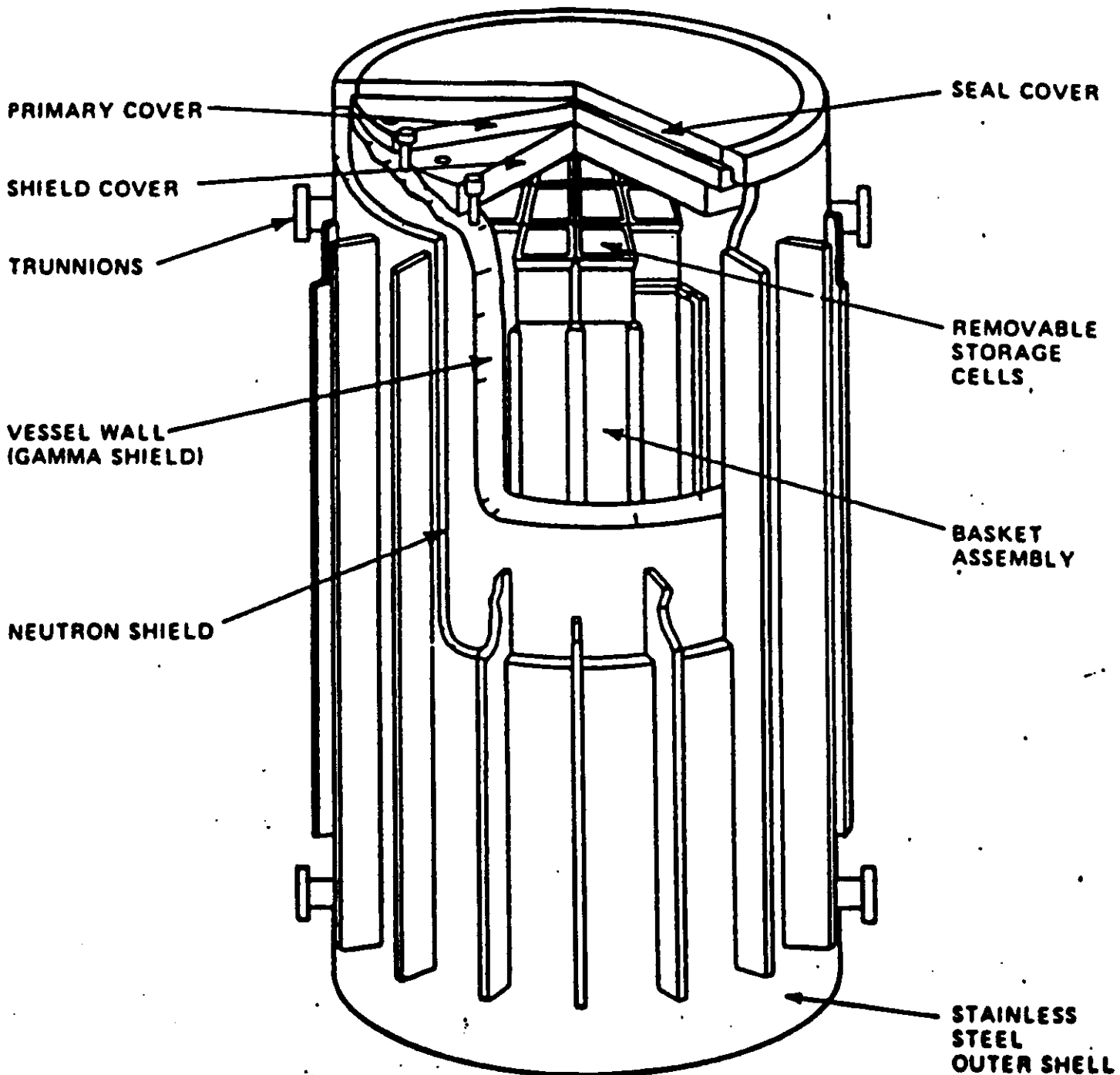
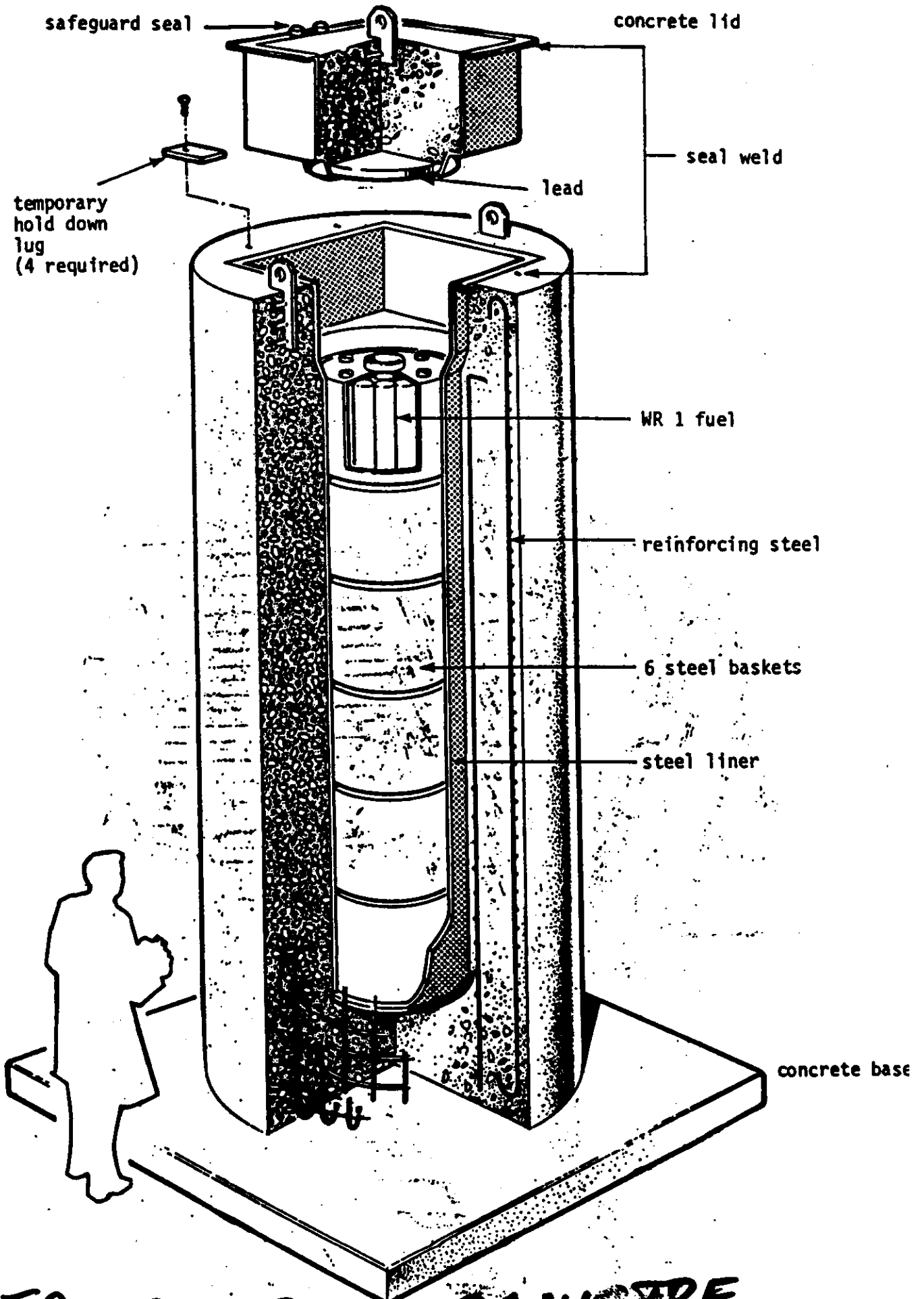


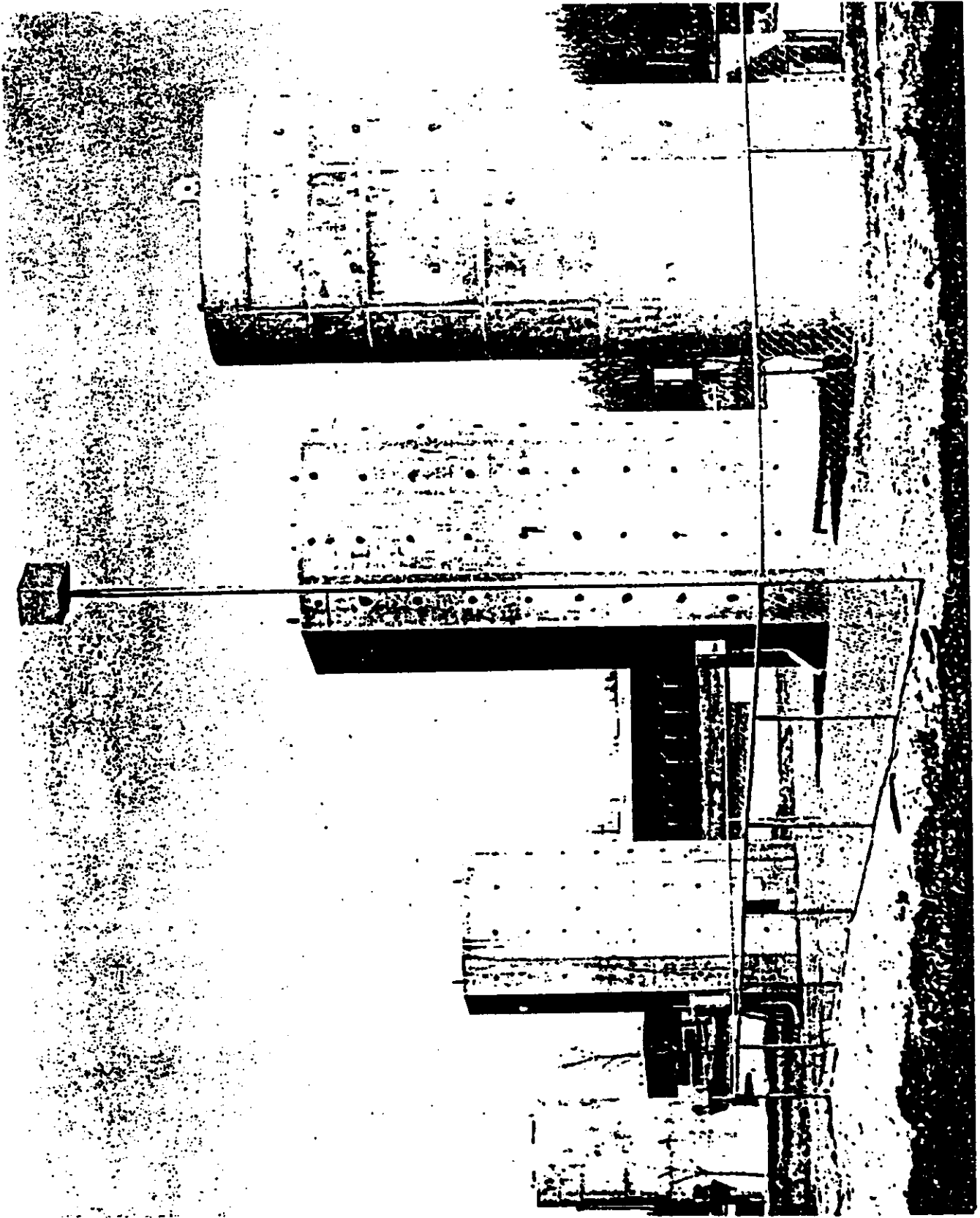
FIGURE 3: CENTRAL POOLS - CUTAWAY (From Ref. 5). AEC-L-6191



e 5-18 Westinghouse MC-10 cask. (Reprinted from *Nuclear News*)



**F2 CONCRETE CANISTER**



AECL 6191

FIGURE 10: CANISTERS IN OPERATION.

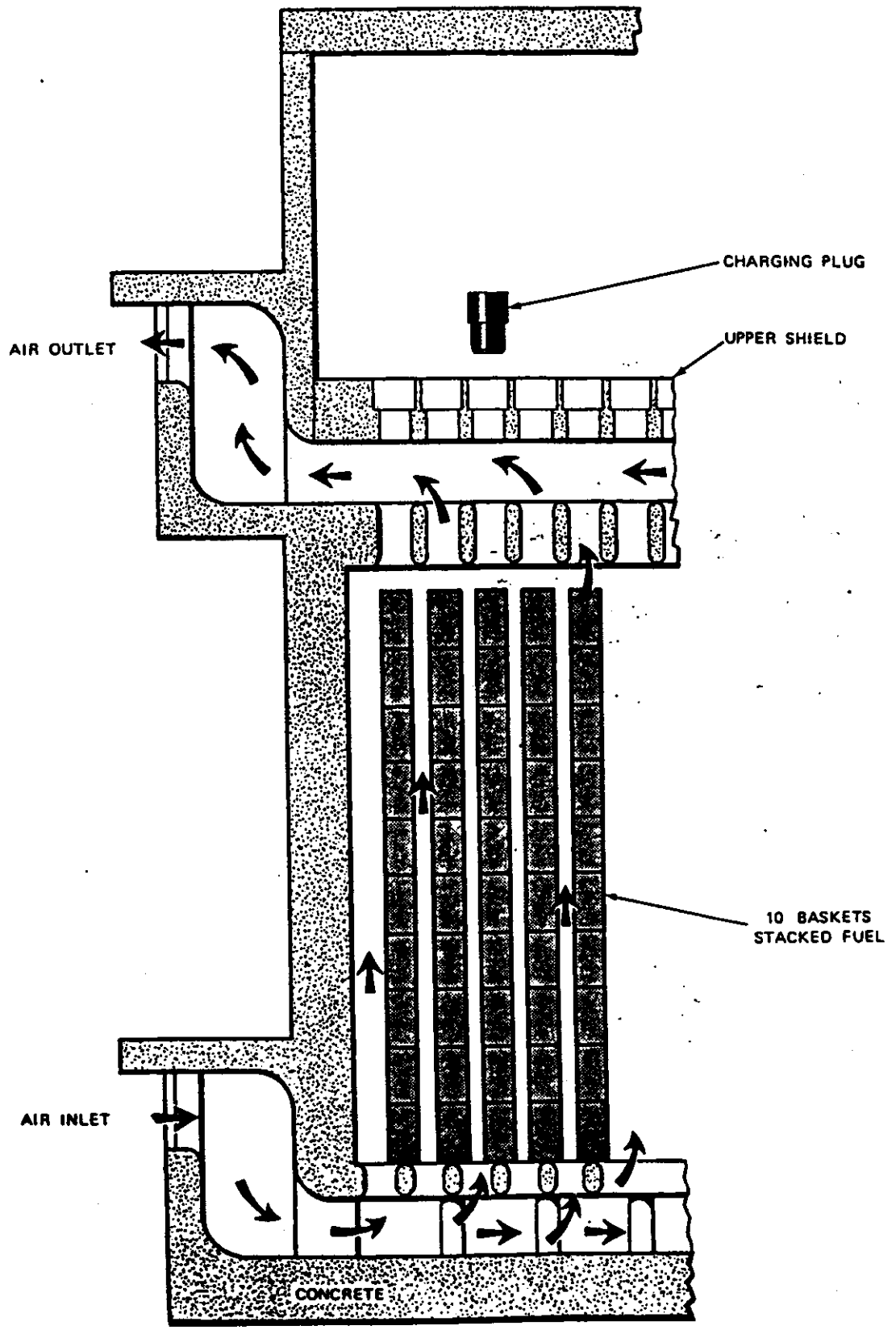


FIGURE 5: DRY STORAGE FACILITY - CONVECTION COOLING.

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